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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of

Richard D. Breault et al

Docket: C-2789

Serial No. 10/668,868

Art Unit: 1745

Filed: September 22, 2003

Examiner: Ben Lewis

Title: Internal PEM Fuel Cell Water Management

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

DECLARATION UNDER 37 CFR 1.132

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

Sir:

I, Richard D. Breault declare that:

1. I reside at 153 Pleasant Street, North Kingstown, RI 02852.
2. I have a Bachelor of Science degree in Chemical Engineering and
have been working in the field of fuel cells and related arts for over 43 years, and am
currently engaged in that field on behalf of UTC Power Corporation, South Windsor,
CT.

3. I am a co-Inventor of the subject application and am currently
familiar with its content, including claims 18 and 19 and the support for those claims
in the specification.

4. The amount of water produced in a fuel cell was widely known, as in
US 6,451,468, issued in 2002, from the reaction at the cathode:



two gram-moles of water molecules are produced for every four gram-moles
of electrons. The Faraday constant, on page 778 of McGraw-Hill Dictionary of
Scientific and Technical Terms, Sixth Edition (Library of Congress data, 2002), copy
herewith, is 96,845 C (coulombs) per each gram-mole of water produced. Since four

gram-moles of electrons are produced for two gram-moles of water, $96,846 \times 2 = 193,690$ coulombs are produced for each mole of water. One coulomb per second is one ampere; thus it is known that 5.1×10^{-6} moles of water are produced per second per ampere of current produced by the fuel cell power plant.

5. The amount of water transferred directly from cathodes to anodes through the water transfer capability or path is controlled (as is explained at page 4, line 24 through page 5, line 16 of the subject application) in response to a pressure differential between the cathode and the anode. The relationship between the pressure differential and the rate of flow (volume per unit of time) of water has been, before September 2003, determined routinely using Darcy's law, described in a contemporaneous printout from Wikipedia, herewith, but also described on page 546 of the 2002 edition of *McGraw-Hill* (hereinbefore), copy herewith.

Since the cathode-to-anode water transfer path will have been designed by, made, and installed by or under the direction of the same skilled artisan who will determine the flow from cathodes to anodes, the permeability, area of flow and length of flow of the water transfer path will be known by that artisan. The viscosity of water around 80°C (for instance) is shown on pages 6-2 and 6-175 of the *Handbook of Chemistry and Physics, 86th Edition* (2006), copy herewith. Viscosity at other temperatures was available from standard tables, such as earlier editions of the *Handbook*. Thus, the establishment of a desired volume of water per unit of time through the water transfer capability or path was a routine matter before September, 2003.

6. The amount of water being discharged from the anodes, through the anode water transport plates, to ambient has routinely been measured by flow meters, well known prior to September, 2003, as shown on page 828 of *McGraw-Hill*, (hereinbefore), copy herewith. This amount is adjusted by varying the cathode-to-anode flow as described in paragraph 8, hereinbefore.

7. The amount of water vapor in the cathode gas flow field exhaust can easily be determined by simply condensing the vapor, as shown in US 5,998,058, issued in 1999, and measuring the flow of water to ambient with a known flow meter.

It was standard knowledge before September, 2003 that the amount of water in a flow of gas, such as the cathode gas flow field exhaust, can also be

determined from the total mass flow, temperature, total pressure, and vapor pressure of water. The vapor pressure of water at a temperature, is found in well-known tables, such as at pages 6-8 of the *Handbook* (hereinbefore) and earlier editions thereof. The mass flow (moles) of water is simply the total flow times the ratio of vapor pressure to total pressure.

8. The creation of flows set forth in either of claim 18 or claim 19 begins by establishing the current to be produced by the stack. From the current, the product water is routinely calculated as described in paragraph 4, hereinbefore. The coolant temperature, and therefore the temperature at the reactant gas exits of the stack is controlled as shown in US 2002/0148608 A1. The pressure of anode and cathode, and thus their relative pressure, is adjustable as shown in US 2001/0004501 A1, to cause the desired cathode-to-anode water flow of the invention, as described in paragraph 5, hereinbefore. The experimentation required by one skilled in the fuel cell arts to balance the cathode exit temperature and the cathode/anode pressure differential, while measuring anode water transport plate outflow and cathode vapor outflow, is routine and easily completed in a matter of hours.

9. All statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true and further that these statements are made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code.

Richard D. Breaux
Richard D. Breaux

December 21, 2007
Date

Darcy's law

From Wikipedia, the free encyclopedia

In fluid dynamics, Darcy's law is a phenomenologically derived constitutive equation that describes the flow of a fluid through a porous medium. The law was formulated by Henry Darcy based on the results of experiments^[1] on the flow of water through beds of sand. It also forms the scientific basis of fluid permeability used in the earth sciences.

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Background

Although Darcy's law (an expression of conservation of momentum) was determined experimentally by Darcy, it has since been derived from the Navier-Stokes equations via homogenization. It is analogous to Fourier's law in the field of heat conduction, Ohm's law in the field of electrical networks, or Fick's law in diffusion theory.

One application of Darcy's law is to water flow through an aquifer. Darcy's law along with the equation of conservation of mass are equivalent to the groundwater flow equation, one of the basic relationships of hydrogeology. Darcy's law is also used to describe oil, water, and gas flows through petroleum reservoirs.

Description

Darcy's law is a simple proportional relationship between the instantaneous discharge rate through a porous medium, the viscosity of the fluid and the pressure drop over a given distance.

$$Q = \frac{-\kappa A (P_b - P_a)}{\mu L}$$

The total discharge, Q (units of volume per time, e.g., m^3/s) is equal to the product of the permeability (κ units of area, e.g. m^2) of the medium, the cross-sectional area (A) to flow, and the pressure drop ($P_b - P_a$), all divided by the dynamic viscosity μ (in SI units e.g. $\text{kg}/(\text{m}\cdot\text{s})$ or $\text{Pa}\cdot\text{s}$), and the length L the pressure drop is taking place over. The negative sign is needed because fluids flows from high pressure to low pressure. So if the change in pressure is PAGE 8/19 *RCVD AT 12/28/2007 1:45:21 PM [Eastern Standard Time]* SVR:USPTO-EFXRF-1/13 * DNIS:2738300 * CSID:8606491385 * DURATION (mm:ss):09:24

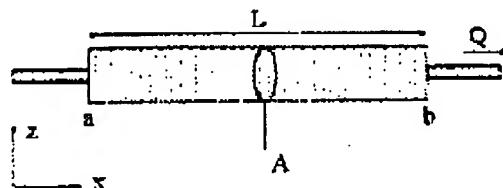


Diagram showing definitions and directions for Darcy's law.

negative (in the x -direction) then the flow will be positive (in the x -direction). Dividing both sides of the equation by the area and using more general notation leads to

$$q = \frac{-k}{\mu} \nabla P$$

where q is the flux (discharge per unit area, with units of length per time, m/s) and ∇P is the pressure gradient vector. This value of flux, often referred to as the Darcy flux, is not the velocity which the water traveling through the pores is experiencing^[2].

The pore velocity (v) is related to the Darcy flux (q) by the porosity (ϕ). The flux is divided by porosity to account for the fact that only a fraction of the total formation volume is available for flow. The pore velocity would be the velocity a conservative tracer would experience if carried by the fluid through the formation.

$$v = \frac{q}{\phi}$$

In 3D

In three dimensions, gravity must be accounted for, as fluid will not flow vertically as a result of the vertical gravitational pressure drop (this is hydrostatic conditions). The correction is to subtract the gravitational pressure drop from the existing pressure drop in the equation in order to express the resulting fluid flow,

$$q = \frac{-K}{\mu} (\nabla P - \rho g \hat{e}_z)$$

where the flux Q is now a vector quantity, K is a tensor of permeability, ∇ is the gradient operator in 3D, g is the acceleration due to gravity, \hat{e}_z is the unit vector in the vertical direction, pointing downwards and ρ is the density.

Effects of anisotropy are addressed in three-dimensions using a symmetric second-order tensor of permeability:

$$K = \begin{bmatrix} K_{xx,x} & K_{xx,y} & K_{xx,z} \\ K_{yy,x} & K_{yy,y} & K_{yy,z} \\ K_{zz,x} & K_{zz,y} & K_{zz,z} \end{bmatrix}$$

where the magnitudes of permeability in the x , y , and z component directions are specified. Since this a symmetric matrix, there are at most six unique values. If the permeability is isotropic (equal magnitude in all directions), then the diagonal values are equal, $K_{xx} = K_{yy} = K_{zz} > 0$, while all other components are 0. The permeability tensor can be interpreted through an evaluation the relative magnitudes in each component. For example, rock with highly permeable vertical fractures aligned in the x -direction will have relatively higher values for K_{xx} than other component values.

Assumptions

Darcy's law is a simple mathematical statement which neatly summarizes several familiar properties that groundwater flowing in aquifers exhibits, including:

- if there is no pressure gradient over a distance, no flow occurs (this is hydrostatic conditions).
- if there is a pressure gradient, flow will occur from high pressure towards low pressure (opposite the

McGRAW-HILL DICTIONARY OF SCIENTIFIC AND TECHNICAL TERMS

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On the cover Representation of a fullerene molecule with a noble gas atom trapped inside. At the Permian-Triassic sedimentary boundary the noble gases helium and argon have been found trapped inside fullerenes. They exhibit isotope ratios quite similar to those found in meteorites, suggesting that a fireball meteorite or asteroid exploded when it hit the Earth, causing major changes in the environment. (Image copyright © Dr. Luann Becker. Reproduced with permission.)

Over the six editions of the Dictionary, material has been drawn from the following references: G. M. Garrity et al., *Taxonomic Outline of the Prokaryotes*, Release 2, Springer-Verlag, January 2002; D. W. Linzey, *Vertebrate Biology*, McGraw-Hill, 2001; J. A. Pechenik, *Biology of the Invertebrates*, 4th ed., McGraw-Hill, 2000; U.S. Air Force *Glossary of Standardized Terms*, AF Manual 11-1, vol. 1, 1972; F. Casey, ed., *Compilation of Terms in Information Sciences Technology*, Federal Council for Science and Technology, 1970; *Communications-Electronics Terminology*, AF Manual 11-1, vol. 3, 1970; P. W. Thrush, comp. and ed., *A Dictionary of Mining, Mineral, and Related Terms*, Bureau of Mines, 1968; *A DOD Glossary of Mapping, Charting and Geodetic Terms*, Department of Defense, 1967; J. M. Gilliland, *Solar-Terrestrial Physics: A Glossary of Terms and Abbreviations*, Royal Aircraft Establishment Technical Report 67158, 1967; W. H. Allen, ed., *Dictionary of Technical Terms for Aerospace Use*, National Aeronautics and Space Administration, 1965; *Glossary of Space Terminology*, Office of Aerospace Research, U.S. Air Force, 1963; *Naval Dictionary of Electronic, Technical, and Imperative Terms*, Bureau of Naval Personnel, 1962; R. E. Huschke, *Glossary of Meteorology*, American Meteorological Society, 1959; *ADP Glossary*, Department of the Navy, NAVSO P-3097; *Glossary of Air Traffic Control Terms*, Federal Aviation Agency; *A Glossary of Range Terminology*, White Sands Missile Range, New Mexico, National Bureau of Standards, AD 467-424; *Nuclear Terms: A Glossary*, 2d ed., Atomic Energy Commission.

**McGRAW-HILL DICTIONARY OF SCIENTIFIC AND TECHNICAL TERMS,
Sixth Edition**

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is the safety limit for a vessel approaching a dangerous obstacle such as a reef. { 'dān-jər ,əf'gəl }

danger area [NAV] A specified area within, below, or over which there may exist activities constituting potential danger to aircraft flying over it, or to persons, property, and traffic on land or sea. { 'dān-jər 'er-əs }

danger bearing [NAV] The bearing of any object or obstruction as measured on board a vessel which will put a ship in jeopardy. { 'dān-jər ,ber'-ɪŋ }

danger buoy [NAV] A buoy marking an isolated danger to navigation. { 'dān-jər ,boi' }

danger coefficient [NUCL.EQ] The change in reactivity per unit mass of a substance resulting from inserting the substance in a particular nuclear reactor. { 'dān-jər ,kō-i,fish-ənt }

danger line [NAV] A line on a chart representing a boundary, beyond which some hazard will be encountered. { 'dān-jər ,lin' }

dangerous semicircle [METEOROL] The half of the circular area of a tropical cyclone having the strongest winds and heaviest seas, where a ship tends to be drawn into the path of the storm. { 'dān-jərəs 'sem-i,sər-kəl' }

danger sounding [NAV] A minimum sounding chosen for a vessel of specific draft in a given area to indicate the limit of safe navigation. { 'dān-jər ,soun-dɪŋ' }

danger space [ORD] 1. That portion of the range within which a target of given dimensions could be hit by a projectile with a given angle of fall. 2. Space around the bursting point of an antiaircraft projectile. { 'dān-jər ,spās' }

dangler [MBT] The flexible electrode used in barrel plating. { 'dāng-glər' }

dangling bond [SOLID STATE] A chemical bond associated with an atom in the surface layer of a solid that does not join the atom with a second atom but extends in the direction of the solid's exterior. { 'dāng-glɪŋ 'bənd' }

dangling ELSE [COMPUT SCI] A situation in which it is not clear to which part of a compound conditional statement an ELSE instruction belongs. { 'dāng-glɪŋ 'els' }

Danian [GEOL] Lowermost Paleocene or uppermost Cretaceous geological time. { 'dān-ē-ən' }

Daniell cell [PHYS CHEM] A primary cell with a constant electromotive force of 1.1 volts, having a copper electrode in a copper sulfate solution and a zinc electrode in dilute sulfuric acid or zinc sulfate, the solutions separated by a porous partition or by gravity. { 'dān-yēl ,sel' }

Daniell hygrometer [ENC] An instrument for measuring dew point; dew forms on the surface of a bulb containing ether which is cooled by evaporation into another bulb, the second bulb being cooled by the evaporation of ether on its outer surface. { 'dān-yēl hī'grām-əd-or' }

Danjon prismatic astrolabe [ENC] A type of astrolabe in which a Wollaston prism just inside the focus of the telescope converts converging beams of light into parallel beams, permitting a great increase in accuracy. { 'dān-yān prīz'māt'ik 'as-trə-ləb' }

dannemorite [MINERAL] $(\text{Fe},\text{Mn},\text{Mg})_3\text{Si}_5\text{O}_{22}(\text{OH})_2$ A yellowish-brown to greenish-gray monoclinic mineral consisting of a columnar or fibrous amphibole. { 'dān-ə-mōr,īt' }

danoyl chloride [ORG CHEM] $(\text{CH}_3)_2\text{NCO}_2\text{Cl}$ A reagent for fluorescent labeling of amines, amino acids, proteins, and phenols. { 'dān-əl 'klōr'īd' }

Danzig reaction [IMMUNOL] A toxin-antitoxin reaction that occurs when an exact equivalence of toxin is added to antitoxin, not in one portion but in successive increments. { 'dā-nish ,rāk-shən' }

DAP See dialyl phthalate; diaminopimelate.

Daphniphyllales [BOT] An order of dicotyledonous plants in the subclass Hamamelidae, consisting of a single family with one genus, *Daphniphyllum*, containing about 35 species; dioecious trees or shrubs native to eastern Asia and the Malay region, they produce a unique type of alkaloid and often accumulate aluminum and sometimes produce iridoid compounds. { 'daf-ni-fil'ə-lēz' }

daphnite [MINERAL] $(\text{Mg},\text{Fe})(\text{Fe},\text{Al}),(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2$ A mineral of the chlorite group consisting of a basic aluminosilicate of magnesium, iron, and aluminum. { 'dāf-nīt' }

Daphoenidae [PALEON] A family of extinct carnivorous mammals in the superfamily Miacoidae. { 'dāf-nē-ōdē' }

dapsone See 4,4'-sulfonyldianiline. { 'dāp-sōn' }

daraf [ELEC] The unit of elastance, equal to the reciprocal of 1 farad. { 'dār-əf' }

darapskite [MINERAL] $\text{Na}_3(\text{NO}_3)(\text{SO}_4)\cdot\text{H}_2\text{O}$ A naturally occurring hydrate mineral consisting of a hydrous nitrate-sulfate of sodium. { 'dār-ap,skīt' }

Darboux's monodromy theorem [MATH] The proposition that, if the function $f(z)$ of the complex variable z is analytic in a domain D bounded by a simple closed curve C , and is continuous in the union of D and C and is injective on C , then $f(z)$ is injective for z in D . { 'dār-būks mōnō-dro'mē-thərm' }

darby [ENO] A flat-surfaced tool for smoothing planes. { 'dār-bē' }

darcy [PHYS] A unit of permeability, equivalent to the sage of 1 cubic centimeter of fluid of 1 centipoise viscosity flowing in 1 second under a pressure of 1 atmosphere through a porous medium having a cross-sectional area of 1 square centimeter and a length of 1 centimeter. { 'dār-sē' }

Darcy number 1 [FL MECH] A dimensionless group, equal to four times the Fanning friction factor. Symbolized D_{ar} . Also known as Darcy-Weisbach coefficient; resistance coefficient 2. { 'dār-sē ,nōm-bər 'wān' }

Darcy number 2 [FL MECH] A dimensionless group, the study of the flow of fluids in porous media, equal to fluid velocity times the flow path divided by the permeability of the medium. Symbolized D_{ar}^2 . { 'dār-sē ,nōm-bər 'wān' }

Darcy's law [FL MECH] The law that the rate at which fluid flows through a permeable substance per unit area is equal to the permeability, which is a property only of the substance through which the fluid is flowing, times the pressure drop per unit length of flow, divided by the viscosity of the fluid. { 'dār-sēz ,lō' }

Darcy-Weisbach coefficient See Darcy number 1. { 'dār-sē ,nōm-bər 'wān' }

Darcy-Weisbach equation [FL MECH] An equation of loss of head due to friction h_f during turbulent flow of a fluid through a duct of any shape; in the case of a circular pipe $f(L/d)(V^2/2g)$, where L and d are the length and diameter of the pipe, V is the fluid velocity, g the acceleration of gravity, and f a dimensionless number called Darcy number 1. { 'dār-sē ,wāi-sāb ,zhōn' }

dark box [GRAPHICS] A light-proof box used to store light-sensitive photographic papers. { 'dārk ,bōks' }

dark cloud [ASTRON] A relatively dense, cool cloud of stellar gas, chiefly molecular, whose dust particles obstruct the light of stars behind it. { 'dārk 'klaud' }

dark conduction [ELECTR] Residual conduction in a sensitive substance that is not illuminated. { 'dārk kō-dənshən' }

dark current See electrode dark current. { 'dārk ,kōr'ēnt' }

dark-current pulse [ELECTR] A phototube dark-current excursion that can be resolved by the system employing phototubes. { 'dārk ,kōr-ēnt' ,pūls' }

dark discharge [ELECTR] An invisible electrical discharge in a gas. { 'dārk 'dis-chärj' }

dark-reciprocating variables [ASTRON] A binary star comprising a bright star and an almost dark companion that revolve about each other. { 'dārk ,rēk-lə-pāt'īng 'vār'ē-bəlz' }

dark-field illumination [OPTICS] A method of illumination in which the illuminating beam is a hollow cone of light formed by an opaque stop at the center of the cone; the specimen is placed at the concentration of the light and is seen with light scattered or diffracted by it. { 'dārk-fīld ,īl-mū-tāshən' }

dark lightning [GRAPHICS] A photographic effect in which lightning gives black photographic streaks instead of white due to multiple exposures caused by successive members of a composite flash. Also known as Claydon effect. { 'dārk 'līt-nēng' }

dark-line spectrum [SPECTR] The absorption spectrum results when white light passes through a substance, consisting of dark lines against a bright background. { 'dārk 'spék-trəm' }

dark matter [ASTRON] Matter that is postulated to explain the rotational motion of the Milky Way and other galaxies, to explain the motions of galaxies in clusters, and, in certain cosmological theories, to achieve the

DARK-FIELD ILLUMINATION

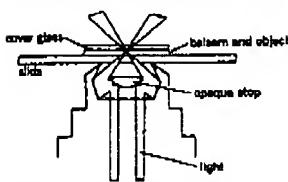


Diagram of dark-field illumination.
(American Optical Corp.)

778 | fan total head

fan total head [MECH ENG] The sum of the fan static head and the velocity head at the fan discharge corresponding to a given quantity of airflow. (/fæn /təl hɛd /hɛd)

fan total pressure [MECH ENG] The algebraic difference between the mean total pressure at the fan outlet and the mean total pressure at the fan inlet. (/fæn /təl p'resər'v)

fan truss [CIV ENG] A truss with struts arranged as radiating lines. (/fæn ,trʌs)

fan vaulting [ARCH] Vaulting in which the ribs diverge like the rays of a fan. (/fæn ,vɔlt'ɪŋ)

fan velocity pressure [MECH ENG] The velocity pressure corresponding to the average velocity at the fan outlet. (/fæn və'lɪsɪ-dəs ,presh'er)

FAQ See Frequently Asked Questions.

farad [ELEC] The unit of capacitance in the meter-kilogram-second system, equal to the capacitance of a capacitor which has a potential difference of 1 volt between its plates when the charge on one of its plates is 1 coulomb, there being an equal and opposite charge on the other plate. Symbolized F. (/fɑ:rad)

→ **Faraday** [PHYS] The electric charge required to liberate 1 gram-equivalent of a substance by electrolysis; experimentally equal to 96,485.3415 ± 0.0039 coulombs. Also known as Faraday constant. (/fɑ:rədɪ)

Faraday birefringence [OPTICS] Difference in the indices of refraction of left and right circularly polarized light passing through matter parallel to an applied magnetic field; it is responsible for the Faraday effect. (/fɑ:rədɪ ,brɪrɪ'njəns)

Faraday cage See Faraday shield. (/fɑ:rədɪ ,kæj)

→ **Faraday constant** See faraday. (/fɑ:rədɪ ,kən'stənt)

Faraday cylinder [ELEC] 1. A closed, or nearly closed, hollow conductor, usually grounded, within which apparatus is placed to shield it from electrical fields. 2. A nearly closed, insulated, hollow conductor, usually shielded by a second grounded cylinder, used to collect and detect a beam of charged particles. (/fɑ:rədɪ ,sɪl'ən dɪs')

Faraday dark space [ELECTR] The relatively nonluminous region that separates the negative glow from the positive column in a cold-cathode glow-discharge tube. (/fɑ:rədɪ ,dɑ:k spes)

Faraday disk machine [ELECTROMAG] A device for demonstrating electromagnetic induction, consisting of a copper disk in which a radial electromotive force is induced when the disk is rotated between the poles of a magnet. Also known as Faraday generator. (/fɑ:rədɪ ,dɪsk mə'hین)

Faraday effect [OPTICAL] Rotation of polarization of a beam of linearly polarized light when it passes through matter in the direction of an applied magnetic field; it is the result of Faraday birefringence. Also known as Faraday rotation; Kundt effect; magnetic rotation. (/fɑ:rədɪ ,dɪ'fekt)

Faraday generator See Faraday disk machine. (/fɑ:rədɪ ,jen'ə,rā'tōr)

Faraday ice bucket experiment [ELEC] Experiment in which one lowers a charged metal body into a pail and observes the effect on an electroroscope attached to the pail, with and without contact between body and pail; the experiment shows that charge resides on a conductor's outside surface. (/fɑ:rədɪ ,ɪs'bak'ət ik,səp'rə'mənt)

Faraday rotation See Faraday effect. (/fɑ:rədɪ ,rə'tā'shən)

Faraday rotation experiment [ELECTROMAG] An experiment in which a wire dipping in a pool of mercury surrounding a magnet rotates around the magnet when current passes through it, demonstrating the effect of a magnetic field on a current-carrying conductor. (/fɑ:rədɪ ,rə'tā'shən ik,səp'rə'mənt)

Faraday rotation isolator See ferrite isolator. (/fɑ:rədɪ ,rə'tā'shən īz'ol,ātōr)

Faraday screen See Faraday shield. (/fɑ:rədɪ ,scren')

Faraday shield [ELEC] Electrostatic shield composed of wire mesh or a series of parallel wires, usually connected at one end to another conductor which is grounded. Also known as Faraday cage; Faraday screen. (/fɑ:rədɪ ,shɪld')

Faraday's law of electromagnetic induction [ELECTROMAG] The law that the electromotive force induced in a circuit by a changing magnetic field is equal to the negative of the rate of change of the magnetic flux linking the circuit. Also known as law of electromagnetic induction. (/fɑ:rədɪ ,lɔ: ɪn'duk'tɪv,magnɪtɪdɪk ,ɪn'duk'shən)

Faraday's laws of electrolysis [PHYS CHEM] 1. The amount

farsightedness

fatravol

of any substance dissolved or deposited in electrical proportion to the total electric charge passed. 2. The different substances dissolved or deposited by the same electric charge are proportional to their weights. (/fɑ:rədɪ ,lɔ:z ɪdəs ,pəsəd')

Faraday tube [ELEC] A tube of force for element which is of such size that the integral over across the tube of the component of electric dipperpendicular to that surface is unity. (/fɑ:rədɪ ,tu:b)

faradio current Also spelled faradai current. electric current that corresponds to the reduction of a chemical species. (/fɑ:rədɪ ,k'ur'ənt)

farmer [ELEC] An intermittent electrical alternating current like that obtained from any winding of an induction coil. (/fɑ:rədɪ ,dɪ)

faradization [BIO PHYS] Use of a faradic current on muscles and nerves. (/fɑ:rədɪ ,zā'shən)

faroy See gllanders. (/fɑ:rəy)

far-end crosstalk [COMMUN] Crosstalk that disturbed circuit in the same direction as desired circuit. (/fɑ:r'end 'krɒstəlk)

farewell buoy See sea buoy. (/fər'wel ,bu:y)

Farey sequence [MATH] The Farey sequence is the increasing sequence, from 0 to 1, of fractions whose denominators is equal to or less than n, with expressed in lowest terms. (/fɑ:rē ,s'kwēns)

far field See Fraunhofer region. (/fɑ:r'fi:d')

farinaceous [BOT] Having a mealy surface texture of meal. (/fɑ:rē ,nes')

farinaceous [FOOD] 1. Containing starch or flour. 2. Having a mealy surface texture that is mealy, soft, and friable, for example or a pelagic cooz. (/fɑ:rē ,nes')

farinaceous [BOT] An order that includes

regarded as orders of the Commelinidae in classification. (/fɑ:rē ,nes')

far-infrared maser [ENG] A gas maser that generates having a wavelength well above 100 micrometers up to the present lower wavelength limit of about 10 micrometers for microwave oscillators. (/fɑ:r'in-f'redɪ ,m'zər)

far-infrared radiation [ELECTROMAG] Infrared wavelengths of which are the longest of those region, about 30–1000 micrometers; requires imaging for spectroscopic analysis. (/fɑ:r'in-f'redɪ ,r'æk'shən)

Farinaceous [BOT] The equivalent name for

farinaceous

farinaceous [AGR] Yielding farina, a fine meal-like matter. (/fɑ:rē ,nes') Covered with a white powder

(/fɑ:rē ,nes')

farm [AGR] A tract of land used for cultivation raising animals. (/fɑ:m')

Farmer dosemeter [NUCLEO] A small ionization chamber with an air wall, used for routine measurement of gamma radiation. (/fɑ:r'mer dō'sim'ētər)

farmer's lung [MED] An acute pulmonary disease caused by the inhalation of spores from moldy hay or manure. (/fɑ:r'merz ,lʌŋ')

farmer's year [CLIMATO] In Great Britain, period starting with the Sunday nearest March 1st. (/fɑ:r'merz ,y'ər)

farming [AGR] The skills and practices of agriculture. (/fɑ:m'ɪŋ)

farmstead [AGR] The whole area that constitutes including its land and buildings. (/fɑ:m'stēd')

farmesol [BIOCHEM] $C_{17}H_{32}OH$ A colorless oil extracted from oils of plants such as citronella, neroli, and tuberoses; it has a delicate floral odor, and is an intermediate step in the biological synthesis of cholesterol in animals and in vertebrates; used in perfumery. (/fɑ:m'ēsəl')

Farnsworth Image dissector tube See image tube. (/fɑ:nzworth 'im'æj di,sektər ,tūb')

far point [OPTICS] The farthest point from which an object is distinctly seen; for a normal eye it is at infinity. Also known as punctum remotum.

far region See Fraunhofer region. (/fɑ:r'rej'ən)

faringtonite [MINERAL] $Mg_3(PO_4)_2$ A colorless, white, or yellow phosphate mineral known only from

from the U.S.A. (/fɑ:rɪng'gənīt')

farightedness See hypermetropia. (/fɑ:rɪd'tēs)

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distribution See flow field. ('flo ,di-stri-byu-shen)

) See subsidence mantle. ('flo ,erth)

equation [FL MECH] Equation for the calculation of

the vapor, liquid) flow through conduits or channels;

an interrelation of fluid properties (such as density

, salinity), environmental conditions (such as temperature

, air pressure), and conduit or channel geometry and conditions

(diameter, cross-sectional shape, or surface roughness).

(flo-zhən)

) The characteristic reproductive structure of a

, particularly if some or all of the parts are brightly

('flau-sr)

sulfur [PHARM] One of three forms of pharmaceu-

, made by sublimation; the other two forms are

sulfur and washed sulfur. Also known as sub-

. ('flau-orz sv 'sal-far)

tin See stannic oxide. ('flau-orz sv 'tin)

) [FL MECH] The velocity and the density of a fluid

of position and time. Also known as flow discri-

. ('flo-sis-tō-feld)

) See strain figure. ('flo ,fig-yer)

) [GEOL] Folding in beds, composed of relatively

beds that assume any shape impressed upon them by

rigid surrounding rocks or by the general stress pattern

formed zone; there are no apparent surfaces of slip.

. ('flo)

) [COMPUT SCI] A directed graph that represents

a program, wherein a node in the graph corresponds to

sequential code and branches correspond to decisions

in the program. ('sys ENG) See signal-flow graph.

('flo)

) [MECH ENG] Structural and mechan-

ical behavior of structures immersed in or conveying fluid

as a result of an interaction between the fluid-dynamic

, the inertia, damping, and elastic forces in the struc-

. ('flo-in-dest v'bri-shən)

) [MIN ENG] A concentration

the fact that liquid films in laminar flow possess a

which is not the same in all depths of the film; by this

smaller particles of ore may be washed off while the

particles accumulate and are intermittently removed.

. ('film kən-kən-trā-shən)

) [MET] A furnace from which molten metal

is cast or drawn. ('flo-īng fər-nəs)

) [PETRO ENG] Pressure at the bottom of an

hole (bottom-hole pressure) during normal oil produc-

. ('flo-hol-presh-ər)

) [PETRO ENG] The slope of

pressure plotted against distance measured for

liquid flow in a continuous-flow gas-lift oil well.

. ('flo-prā-grād-ē-ənt)

) [THERMO] Calculation correc-

for gases flowing at temperatures other than that

the flow equation is valid, that is, other than 60°F

(15°-in 'tem-prā-char-fak-tor)

) [PETRO ENG] Oil reservoir in which gas-drive

is sufficient to force oil flow up through and out of

. ('flo-in-wel)

) [CYTOL] A karyotype that is based on flow

measurements. ('flo-kär-ē-tip)

) [PETR] An igneous rock, a layer which is differ-

in composition or texture from adjacent layers. ('flo)

) [ENG] 1. The connecting line or arrow between

the flow chart or block diagram. 2. Mark on a

article made by the meeting of two

during molding. Also known as weld line;

) [HYD] A contour of the water level around a

. (PETR) In an igneous rock, any internal struc-

by parallel orientation of crystals, mineral

. (PETRO ENG) A pipeline that takes oil

the wall or a series of wells to a gathering center.

) [MATER] Wavy surface marks on a thermoplastic

due to improper flow of material into the mold.

. ('flo)

) [ENG] The determination of the quantity

either a liquid, a vapor, or a gas, that passes through

or open channel. ('flo ,meas-or-mont)

flowmeter [ENG] An instrument used to measure pressure,

flow rate, and discharge rate of a liquid, vapor, or gas flowing

in a pipe. Also known as fluid meter. ('flo-med-ər)

) flow mixer [MECH ENG] Liquid-liquid mixing device in

which the mixing action occurs as the liquids pass through it;

includes jet nozzles and agitator vanes. Also known as fine

mixer. ('flo-mik-sər)

) flow net [FL MECH] A diagram used in studying the flow of

a fluid through a permeable substance (such as water through

a soil structure) having two nests of curves, one representing

the flow lines, which follow the path of the fluid, and the other

the equipotential lines, which connect points of equal head.

('flo-net)

) flow noise [ACOUS] 1. Pressure variations associated with

a turbulent flow field that do not propagate away from the

turbulent source but are sensed as sound by a receiver in direct

contact or close to the turbulent flow. Also known as near-field

noise. 2. More generally, any noise generated by turbulent

fluid flow. ('flo-nōɪz)

) flow nozzle [ENG] A flowmeter in a closed conduit, consisting of a short flared nozzle of reduced diameter inset into

the inner diameter of a pipe; used to cause a temporary pressure

drop in flowing fluid to determine flow rate via measurement

of static pressures before and after the nozzle. ('flo-nōz-əl)

) flow of variability [GEN] The movement of genetic variabil-

ity within a population as a result of hybridization and segreg-

ation. ('flo ev ,ver-e-əbil-i-tē)

) flow pattern [FL MECH] Pattern of two-phase flow in a con-

duit or channel pipe, taking into consideration the ratio of gas

to liquid and conditions of flow resistance and liquid holdup.

('flo-pad-əm)

) flow process [ENG] System in which fluids or solids are

handled in continuous movement during chemical or physical

processing or manufacturing. ('flo-prä-ses)

) flow-programmed chromatography [ANALY CHEM] A

chromatographic procedure in which the rate of flow of the

mobile phase is periodically changed. ('flo-prägram ,krō-ma'tik-ro-fə)

) flow rate [FL MECH] Also known as rate of flow. 1. Time

required for a given quantity of flowable material to flow a

measured distance. 2. Weight or volume of flowable material

flowing per unit time. ('flo-rat)

) flow-rating pressure [MECH ENG] The value of inlet static

pressure at which the relieving capacity of a pressure-relief

device is established. ('flo-rād-in ,presh-ər)

) flow reactor [CHEM ENG] A dynamic reactor system in

which reactants flow continuously into the vessel and products

are continuously removed, in contrast to a batch reactor. ('flo-re-ak-tör)

) flow regime [HYD] A range of streamflows having similar

bed forms, flow resistance, and means of transporting sediment.

('flo-rēzhəm)

) flow resistance [FL MECH] 1. Any factor within a conduit

or channel that impedes the flow of fluid, such as surface

roughness or sudden bends, contractions, or expansions. 2.

See viscosity. ('flo ri-zist-əns)

) flow rock [PETR] An igneous rock that had been liquid.

('flo-rak)

) flow separation See boundary-layer separation. ('flo-sep-ə-rā-shən)

) flow sheet See flow chart. ('flo-shet)

) flow shop [IND ENG] A manufacturing facility in which

machine tools and robots are employed in the same manner on

all jobs. ('flo-shap)

) flow slide [GEOL] A slide of waterlogged material in which

the slip surface is not well defined. ('flo-slīd)

) flow soldering [ENG] Soldering of printed circuit boards by

moving them over a flowing wave of molten solder in a solder

bath; the process permits precise control of the depth of immersion

in the molten solder and minimizes heating of the board.

Also known as wave soldering. ('flo-sold-ə-ring)

) flowstone [GEO] Deposits of calcium carbonate that accu-

mulated against the walls of a cave where water flowed on the

rock. ('flo-stən)

) flow stress [MECH] The stress along one axis at a given

value of strain that is required to produce plastic deformation.

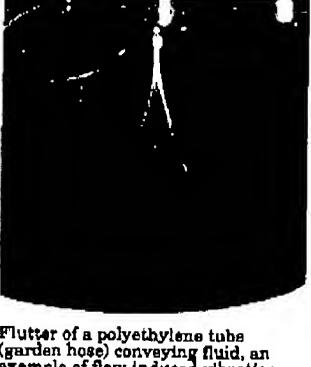
('flo-stres)

) flow string [PETRO ENG] Total length of oil- or gas-well

flow string

825

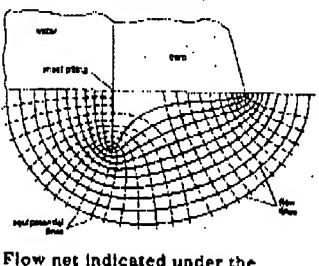
FLOW-INDUCED VIBRATION



Flutter of a polyethylene tube (garden hose) conveying fluid, an example of flow-induced vibration.

The tube is illuminated by a stroboscope to show its motion.

FLOW NET



Flow net indicated under the cutoff wall of a dam. (From D. P. Krymke, Soil Mechanics, 3d ed., McGraw-Hill, 1947)

FLOW NOZZLE

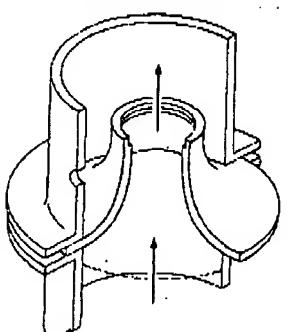
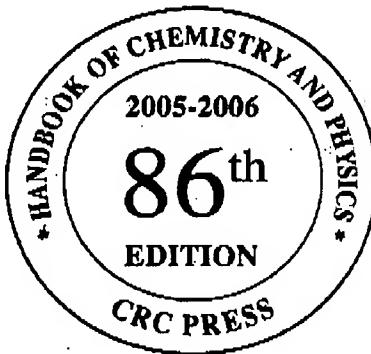


Diagram of a flow nozzle. Arrows indicate direction of fluid flow.

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PROPERTIES OF WATER IN THE RANGE 0-100 °C

This table summarizes the best available values of the density, specific heat capacity at constant pressure (C_p), vapor pressure, viscosity, thermal conductivity, dielectric constant, and surface tension for liquid water in the range 0 - 100 °C. All values (except vapor pressure) refer to a pressure of 100 kPa (1 bar). The temperature scale is IPTS-68.

References

1. L. Harr, J. S. Gallagher, and G. S. Kell, *NBS/NRC Steam Tables*, Hemisphere Publishing Corp., New York, 1984.

2. K. N. Marsh, Ed., *Recommended Reference Materials for the Physicochemical Properties*, Blackwell Scientific Press, Oxford, 1987.
3. J. V. Sengers and J. T. R. Watson, Improved international fits for the viscosity and thermal conductivity of water substances, *Cham. Ref. Data.*, 15, 1291, 1986.
4. D. G. Archer and P. Wang, The dielectric constant of water: Hückel limiting law slopes, *J. Phys. Chem. Ref. Data.*, 19, 9, 1990.
5. N. B. Vargaftik, et al., International tables of the surface tension of water, *J. Phys. Chem. Ref. Data.*, 12, 817, 1983.

t °C	Density g/cm³	C_p J/g K	Vap. pres. kPa	Visc. μPa s	Ther. cond. mW/K m	Diel. const.	Surf. ten. mN/m
0	0.99984	4.2176	0.6113	1793	561.0	87.90	75.64
10	0.99970	4.1921	1.2281	1307	580.0	83.96	74.23
20	0.99821	4.1818	2.3388	1002	598.4	80.20	72.75
30	0.99565	4.1784	4.2455	797.7	615.4	76.60	71.20
40	0.99222	4.1785	7.3814	653.2	630.5	73.17	69.60
50	0.98803	4.1806	12.344	547.0	643.5	69.88	67.94
60	0.98320	4.1843	19.932	466.5	654.3	66.73	66.24
70	0.97778	4.1895	31.176	404.0	663.1	63.73	64.47
80	0.97182	4.1963	47.373	354.4	670.0	60.86	62.67
90	0.96535	4.2050	70.117	314.5	675.3	58.12	60.82
100	0.95840	4.2159	101.325	281.8	679.1	55.51	58.91
Ref.	1-3	2	1, 3	3	3	4	5

ENTHALPY OF VAPORIZATION OF WATER

The enthalpy (heat) of vaporization of water is tabulated as a function of temperature on the IPTS-68 scale.

Reference

Marsh, K. N., Ed., *Recommended Reference Materials for the Physicochemical Properties*, Blackwell, Oxford, 1987.

t °C	$\Delta_{\text{vap}}H$ kJ/mol	t °C	$\Delta_{\text{vap}}H$ kJ/mol
0	45.054	200	34.962
25	43.990	220	33.468
40	43.350	240	31.809
60	42.482	260	29.930
80	41.585	280	27.795
100	40.657	300	25.800
120	39.684	320	22.297
140	38.643	340	18.502
160	37.518	360	12.966
180	36.304	374	2.066

VAPOR PRESSURE OF WATER FROM 0 TO 370°C

This table gives the vapor pressure of water at intervals of 1°C from the melting point to the critical point.

Reference

Haar, L., Gallagher, J. S., and Kell, G. S., NBS/NRC
Hemisphere Publishing Corp., New York, 1984.

<i>t</i> /°C	<i>P</i> /kPa	<i>t</i> /°C	<i>P</i> /kPa	<i>t</i> /°C	<i>P</i> /kPa	<i>t</i> /°C
0	0.61129	52	13.623	104	116.67	156
1	0.65716	53	14.303	105	120.79	157
2	0.70605	54	15.012	106	125.03	158
3	0.75813	55	15.752	107	129.39	159
4	0.81359	56	16.522	108	133.88	160
5	0.87260	57	17.324	109	138.50	161
6	0.93537	58	18.159	110	143.24	162
7	1.00211	59	19.028	111	148.12	163
8	1.0730	60	19.932	112	153.13	164
9	1.1482	61	20.873	113	158.29	165
10	1.2281	62	21.851	114	163.58	166
11	1.3129	63	22.868	115	169.02	167
12	1.4027	64	23.925	116	174.61	168
13	1.4979	65	25.022	117	180.34	169
14	1.5988	66	26.163	118	186.23	170
15	1.7056	67	27.347	119	192.28	171
16	1.8185	68	28.576	120	198.48	172
17	1.9380	69	29.852	121	204.85	173
18	2.0644	70	31.176	122	211.38	174
19	2.1978	71	32.549	123	218.09	175
20	2.3388	72	33.972	124	224.96	176
21	2.4877	73	35.448	125	232.01	177
22	2.6447	74	36.978	126	239.24	178
23	2.8104	75	38.563	127	246.66	179
24	2.9860	76	40.205	128	254.25	180
25	3.1690	77	41.905	129	262.04	181
26	3.3629	78	43.665	130	270.02	182
27	3.5670	79	45.487	131	278.20	183
28	3.7818	80	47.379	132	286.57	184
29	4.0078	81	49.324	133	295.15	185
30	4.2455	82	51.342	134	303.93	186
31	4.4953	83	53.428	135	312.98	187
32	4.7578	84	55.585	136	322.14	188
33	5.0395	85	57.815	137	331.57	189
34	5.3229	86	60.119	138	341.22	190
35	5.6267	87	62.499	139	351.09	191
36	5.9453	88	64.958	140	361.19	192
37	6.2795	89	67.496	141	371.53	193
38	6.6298	90	70.117	142	382.11	194
39	6.9969	91	72.823	143	392.92	195
40	7.3814	92	75.614	144	403.98	196
41	7.7840	93	78.494	145	415.29	197
42	8.2054	94	81.465	146	426.85	198
43	8.6463	95	84.529	147	438.67	199
44	9.1075	96	87.688	148	450.75	200
45	9.5898	97	90.945	149	463.10	201
46	10.094	98	94.301	150	475.72	202
47	10.620	99	97.759	151	488.61	203
48	11.171	100	101.32	152	501.78	204
49	11.745	101	104.99	153	515.23	205
50	12.344	102	108.77	154	528.96	206
51	12.970	103	112.66	155	542.99	207

VISCOSITY OF LIQUIDS

x-values for viscosity of some common liquids at temperatures up to 100°C is given in this table. Values were derived by fitting experimental data to suitable expressions for the temperature dependence. The substances are arranged by molecular weight in modified Hill order (see Preface). All values are in millipascals seconds (mPa s); this unit is identical

to centipoise; it corresponds to a nominal pressure of 1 atmosphere and a temperature above the normal boiling point. The vapor pressure is understood to be the vapor pressure of the pure liquid at that temperature. A few values are given at a temperature below the normal freezing point; these refer to the pure liquid.

The accuracy of the values given in this table ranges from 1% in the best cases to 5 to 10% in the worst. Additional significant figures are included in the table for interpolation.

References

- Viswanath, D. S. and Natarajan, G., *Data Book on the Viscosity of Liquids*, Hemisphere Publishing Corp., New York, 1989.
- Daubert, T. E., Danner, R. P., Sibul, H. M., and Stebbins, C. C., *Physical and Thermodynamic Properties of Pure Compounds: Data Compilation, extant 1994* (core with 4 supplements), Taylor & Francis, Bristol, PA (also available as database).
- Ho, C. Y., Ed., *CINDAS Data Series on Material Properties*, Vol. V-1, *Properties of Inorganic and Organic Fluids*, Hemisphere Publishing Corp., New York, 1988.
- Stephan, K. and Lucas, K., *Viscosity of Dense Fluids*, Plenum Press, New York, 1979.
- Vargaftik, N. B., *Tables of Thermophysical Properties of Liquids and Gases*, 2nd ed., John Wiley, New York, 1975.

Name	-25°C	0°C	25°C	50°C	75°C	100°C
<i>not containing carbon</i>						
Bromine		1.252	0.944	0.746		
Trichlorosilane		0.415	0.326			
Phosphorous trichloride	0.870	0.662	0.529	0.439		
Tetrachlorosilane			99.4	96.2		
Water		1.793	0.890	0.547	0.378	0.282
Hydrazine				0.876	0.628	0.480
Mercury				1.526	1.402	1.312
Nitrogen dioxide		0.532	0.402			1.245
<i>containing carbon</i>						
Trichlorofluoromethane	0.740	0.539	0.421			
Tetrachloromethane		1.321	0.908	0.656	0.494	
Carbon disulfide		0.429	0.352			
Tribromomethane			1.857	1.367	1.029	
Trichloromethane	0.988	0.706	0.537	0.427		
Hydrogen cyanide		0.235	0.183			
Dibromomethane	1.948	1.320	0.980	0.779	0.652	
Dichloromethane	0.727	0.533	0.413			
Formic acid			1.607	1.030	0.724	0.545
Iodomethane		0.594	0.469			
Formamide		7.114	3.343	1.833		
Nitromethane	1.311	0.875	0.630	0.481	0.383	0.317
Methanol	1.258	0.793	0.544			
Methylamine	0.319	0.231				
1,1,2-Trichlorotrifluoro-ethane	1.465	0.945	0.656	0.481		
Tetrachloroethylene		1.114	0.844	0.663	0.535	0.442
Trichloroethylene		0.703	0.545	0.444	0.376	
Pentachloroethane		3.761	2.254	1.491	1.061	
Trifluoroacetic acid			0.808	0.571		
cis-1,2-Dichloroethylene	0.786	0.575	0.445			
trans-1,2-Dichloroethylene	0.522	0.398	0.317	0.261		
1,1,1,2-Tetrachloroethane	3.660	2.200	1.437	1.006	0.741	0.570
1-Chloro-1,1-difluoro-ethane	0.477	0.376				
Acetyl chloride			0.368	0.294		
1,1,1-Trichloroethane	1.847	1.161	0.793	0.578	0.428	
Acetonitrile		0.400	0.369	0.284	0.234	
1,2-Dibromoethane			1.595	1.116	0.837	0.661